

amine group-containing alkoxysilane having active hydrogen therein (a-2), (b) an acid catalyst, (c) an alkali-soluble UV absorber, (d) at least one organic solvent having a boiling point of from 100 to 250°C, and (e) a dye and/or a pigment.

11. (New) The colored, transparent film-forming composition according to Claim 1, wherein the amino group-containing alkoxysilane having active hydrogen (a-2) is one or more members selected from the group consisting of N-(β-aminoethyl)-γ-aminopropyltrimethoxysilane and N-(β-aminoethyl)-γ-aminopropylmethyldimethoxysilane.

12. (New) The colored, transparent film-forming composition according to Claim 11, wherein the epoxy group-containing alkoxysilane (a-1) is one or more members selected from the group consisting of γ-glycidoxypropyltrimethoxysilane and γ-glycidoxypropylmethyldimethoxysilane.

#### REMARKS

The abstract and specification have been amended in order to correct grammatical and idiomatic errors contained therein. No new matter has been added.

In response to the restriction requirement outlined in the office action, Applicants have canceled Claims 7 and 8 without prejudice to the filing of a divisional application thereon. Claims 1-3 have been amended in order to respond to the Examiner's rejection of Claim 2 under 35 USC 112, second paragraph. Newly presented Claims 9-12 are directed to the preferred embodiments of the present invention. No new matter has been added. It is respectfully submitted that the currently claimed invention is clearly patentably distinguishable over the prior art cited by the Examiner.

Claims 1 and 3-6 have been rejected under 35 USC 102(b) as being anticipated by Nambu et al. Claim 2 has been

rejected under 35 USC 103(a) as being unpatentable over Nambu et al and further in view of Famesh. Applicants respectfully traverse these grounds of rejection and urge reconsideration in light of the following comments.

The presently claimed invention is directed to a colored, transparent film-forming composition which comprises (a) a reaction product of an epoxy group-containing alkoxysilane (a-1) and an amino group-containing alkoxysilane having active hydrogen therein (a-2), (b) an acid catalyst, (c) an alkali-soluble UV absorber, (d) at least one organic solvent having a boiling point of from 100 to 250°C, and (e) a dye and/or a pigment.

As discussed in the present specification, the present invention provides a colored, transparent film-forming composition which is easy to handle, yields an attractive, fashionable, transparent colored film article and can be easily removed or separated from a substrate. The film-forming composition of the present invention has a good leveling property without causing coating irregularities and can be cured at room temperature to provide a film that is free of color shading, high in film strength after curing and readily removable or separated from a substrate after extended use thereof. It is respectfully submitted that the prior art cited by the Examiner does not disclose the presently claimed invention.

The Nambu et al reference discloses a top coat clear coating which contains a composition comprising a hydroxyl group-containing resin, a hydrolyzable silyl group-containing polymer and a hardening catalyst as the main component. Although this reference does disclose that an epoxy group-containing alkoxysilane can be contained in the top coat clear coating as a hardening catalyst (C), the amino group-containing alkoxysilane having active hydrogen contained therein is only disclosed as being an optional additive, this reference has no specific disclosure or examples of a

composition falling within the scope of Claim 1. Therefore, the Examiner's rejection of Claim 1 and 3-6 under 35 USC 101(b) as being anticipated by Nambu et al clearly is in error.

The polymer used in the present invention is formed from an epoxy group-containing alkoxysilane monomer and an amino group-containing alkoxysilane monomer. In contrast thereto, in Nambu et al, the polymer contains a hydroxy group-containing acryl monomer as an indispensable component and includes a monomer other than a silicon-based monomer. As such, the polymer of Nambu is expressly different from that of the present invention in chemical constitution.

The film-forming composition of the present invention can be applied to glass to obtain a uniform film. In contrast thereto, the Nambu et al film forming composition does not have a high leveling property as does the present invention. Moreover, the film forming composition of the present invention can cure in a very short period of time such as 30-40 minutes in the summer and 80-90 minutes in the winter, to form a cured film. The surface characteristics of the film of Nambu et al, after it has dried, is different from that of the present invention with respect to adhesiveness relative to glass and hardness. Therefore, even if other constituents such as a condensation catalyst, an acid catalyst, an alkali-soluble UV absorber, a solvent and a dye and/or a pigment of the present invention was added to the basic composition of Nambu et al, the film forming composition of Nambu et al would be completely different from that of the present invention due to the reaction product required for forming the film-forming composition of the present invention.

The Famesh reference discloses a cross-linked composition formed from combining an isocyanurate and a dicarboxylic acid cross-linking agent. Although this reference discloses that a polyfunctional epoxy compound and a dicarboxylic acid serving as a cross linker can be mixed at a specific ratio to form the

film-forming composition, the film-forming composition of Famesh is neither silicon-based nor corresponds to any epoxy group-containing alkoxysilane of the present invention. Additionally, the amino group-containing alkoxysilane having active hydrogen therein that is an essential component of the film-forming polymer of the present invention is not shown in Famesh. As such, Applicants respectfully submit that the Famesh reference cannot be properly combined with the primary Nambu et al reference to disclose the presently claimed invention as is required under 35 USC 103.

Although the Examiner has not made a showing of prima facie obviousness under 35 USC 103, objective evidence is present in the instant specification which is more than sufficient to rebut any proper showing of prima facie obviousness under 35 USC 103. In Tables 1-3 of the present specification, test results are presented in which film-forming compositions according to the present invention are tested against a comparative film-forming composition. As can be seen by the results in the Tables, the film-forming compositions of the present invention clearly are superior to the reference film-forming composition. Nothing in the reference cited by the Examiner suggests that this would occur. As such, it is respectfully submitted that the patentability of the presently claimed invention over the prior art clearly has been established. Moreover, newly presented Claims 9 and 10 expressly exclude the critical components of the Nambu et al reference therefrom. As such, these claims are even further distinguished over the prior art cited by the Examiner.

The Examiner is respectfully requested to reconsider the present application and to pass it to issue.

Respectfully submitted,

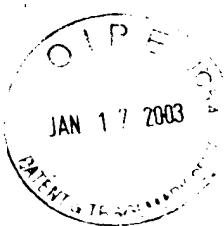
  
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COLORED, TRANSPARENT FILM-FORMING COMPOSITION,  
COATING METHOD AND REMOVAL METHOD OF A FILM THEREOF

ABSTRACT

A colored, transparent film-forming composition, which ~~comprises~~ is made up of (a) a reaction product of an epoxy group-containing alkoxysilane (a-1) and an amino group-containing alkoxysilane (a-2) having active hydrogen therein, (b) an acid catalyst, (c) an alkali-soluble UV absorber, (d) at least one solvent selected from organic solvents having a boiling point of 100 to 250°C, and (e) a dye and/or a pigment, has a good coating performance, ~~a good~~ and room temperature curing characteristic and can provide a film having good film strength and ~~good~~ film removability after use. The coating method of the composition and the removing method of the film obtained from the composition are also described.

IN THE SPECIFICATION

Please amend the specification as follows.

SPECIFICATION

COLORLED, TRANSPARENT FILM-FORMING COMPOSITION,  
~~ITS~~COATING METHOD AND REMOVING~~INGAL~~ METHOD OF A FILM THEREOF

BACKGROUND OF THE INVENTION

Field of the Invention:

This invention relates to a colored, transparent film-forming composition, which is applied onto a transparent substrate made, for example, of glass, plastics or the like and used as a window of buildings or a window of vehicles, including automobiles, thereby forming a uniform, clear, transparent colored film thereon. The invention also relates to a coating method of the composition and a method for removing or separating the resultant film. More particularly, the invention relates to a film-forming composition that is able to form a fashionable film on window glass of automotive vehicles and simultaneously serves to intercept UV light, and ~~its~~a coating method and a ~~removing~~removal method of a film obtained from the composition.

Description of the Prior Art

It is a usual practice to attach, on the window of buildings or automotive vehicles, a plastic film that is formulated with a UV absorber mainly for the purpose of absorbing UV light, or a coating formulated with a UV absorber.

However, the plastic film formulated with a UV absorber has the problems ~~of~~ the kind of color, the

technique of attaching the film, and costs. In addition, further problems are involved in that it is troublesome to cut the film in a desired form with a difficulty in attaching to a complicated portion, air cells are apt to remain upon attachment of the film, and an adhesive is left after removal of the film, so that it has not been satisfactory for everyone to make ready use of it.

Where a film-forming composition containing a UV absorber is used, it is difficult to cure an applied film at room temperature, aside from color shading ascribed to the unevenness of coating, coupled with another deficiency that film strength after curing is low.

Further, a coated film, which is time-worn, has the vital drawback that it is difficult to remove or separate the coated film, and thus, a transparent substrate is very prone to be damaged on the surface thereof.

#### SUMMARY OF THE INVENTION

It is accordingly an object of the invention to provide a colored, transparent film-forming composition, which is easy into handling and can yield an attractive, fashionable, transparent, colored film article and the film can be simply removed or separated off from a substrate.

It is another object of the invention to provide a method for coating such a composition as mentioned above and also a removing method of a film obtained from the composition.

It is a further object of the invention to provide a colored, transparent, film-forming composition, which has a good leveling property without causing coating irregularities and can be cured at room temperature, can provide a film that is free of color shading, is high in



film strength after curing, and is readily removable or separable after having been ~~much~~ used.

It is still a further object of the invention to provide a coating method of the composition mentioned above and also a removingingal method of the film obtained from the composition.

According to the invention, there is provided a colored transparent film-forming composition which comprises (a) a reaction product of an epoxy group-containing alkoxysilane (a-1) and an amino group-containing alkoxysilane ~~(a-2)~~ having active hydrogen therein (a-2), (b) an acid catalyst, (c) an alkali-soluble UV absorber, (d) at least one solvent selected from organic solvents having a boiling point of 100 to 250°C, and (e) a dye and/or a pigment. We have found that this composition has a good coating property, a good room temperature-curing characteristic, a good film strength characteristic, and good film peelability.

#### PREFERRED EMBODIMENTS OF THE INVENTION

The component (a) used in the invention is a reaction product of an epoxy group-containing alkoxysilane (a-1) and an amino group-containing alkoxysilane ~~(a-2)~~ having active hydrogen (a-2) therein. The epoxy group-containing alkoxysilane (a-1) that is one of the constituents of the component (a) includes, for example,  $\gamma$ -glycidoxypropyltrimethoxysilane,  $\gamma$ -glycidoxypropylmethyldimethoxysilane or the like.

The amino group-containing alkoxysilane ~~(a-2)~~ having active hydrogen (a-2), which is the other constituent of the component (a) includes, for example,  $\gamma$ -aminopropyltriethoxysilane, N-( $\beta$ -aminoethyl)- $\gamma$ -aminopropyltrimethoxysilane, N-( $\beta$ -aminoethyl)- $\gamma$ -

aminopropylmethyldimethoxysilane and the like, of which N-( $\beta$ -aminoethyl)- $\gamma$ -aminopropyltrimethoxysilane of the formula,  $\text{H}_2\text{N}(\text{CH}_2)_2\text{NH}(\text{CH}_2)_3\text{Si}(\text{OMe})_3$ , is preferably used.

When N-( $\beta$ -aminoethyl)- $\gamma$ -aminopropyltrimethoxysilane of the formula,  $\text{H}_2\text{N}(\text{CH}_2)_2\text{NH}(\text{CH}_2)_3\text{Si}(\text{OMe})_3$ , exists as a constituent of the component (a), a hard film can be formed after curing and is suitable for application to automotive window glass. If the window is opened or closed, the film is unlikely to suffer scratches therein.

For the reaction between the constituents (a-1) and (a-2), the mixing ratio by weight between the epoxy group-containing alkoxysilane (a-1) and the amino group-containing alkoxysilane ~~(a-2)~~ having active hydrogen (a-2) should preferably be 6:4 to 9:1.

More preferably, the ratio between (a-1) and (a-2) is in the range of 7:3 to 8:2.

When the ratio by weight of the epoxy group-containing alkoxysilane is larger than 9, curing of the resultant colored, transparent film-forming composition after coating proceeds slowly, with the surface hardness of the resultant film becoming low. On the ~~contrary~~ other hand, when the ratio by weight of the amino group-containing alkoxysilane having active hydrogen is larger than 4, the film of the resulting colored-transparent film-forming composition exhibits poor weatherability.

The acid catalyst used in the practice of the invention may be any ones, which acts as a catalyst when the hydroxyl group-containing hydrophilic alkoxysilane compound is hydrolyzed into a more reactive silanol at room temperature and subsequently undergoes polycondensation, -e.g. sulfuric acid, nitric acid, phosphoric acid, para-toluenesulfonic acid and the like. Preferably, boron trifluoride is used.

The alkali-soluble UV absorber useful in the invention may be ones selected from benzophenone compounds, benzotriazole compounds, and the like provided that they are alkali-soluble. Specific examples include 2,4-dihydroxybenzophenone, 2,2', 4,4'-tetrahydroxybenzophenone, 2-hydroxy-4-methoxybenzophenone-5-sulfonic acid and the like.

The alkali-soluble UV absorber is present in an amount of 5 wt% to 40 wt% of the colored, transparent film-forming composition. If the amount of the UV absorber is less than 5 wt%, the UV absorbing effect is lessened along with poor film separability. If the UV absorber exceeds 40 wt%, not only waterproofing and a resistance to chemicals such as an oil film cleaner and the like) are worsened, but also blooming undesirably occurs. From the standpoint of practical use, the amount preferably ranges from 15 wt% to 30 wt%.

The solvent used in the invention is made fundamentally of an alcohol solvent, a cellosolve solvent and a solvent having two or more functional groups, which, respectively, have a boiling point of 100 to 250°C. Typical examples include isobutyl alcohol, diethylene glycol diethyl ether, diisobutyl ketone, propylene glycol monoethyl ether acetate, dipropylene glycol monomethyl ether propanol and the like.

The propylene glycol monoethyl ether acetate includes 1-ethoxy-2-propyl acetate and 2-ethoxy-1-propylacetate, and it is preferred to use a mixture of not less than 90% of 1-ethoxy-2-propylacetate and not larger than 10% of 2-ethoxy-1-propylacetate.

The solvents used in the invention have a boiling point of 100 to 250°C. If the boiling point is lower than 100°C, the leveling property is worsened on coating,

resulting in a poor film appearance. In contrast, when the boiling point exceeds 250°C, a coated film is unlikely to be dried.

Moreover, methyl alcohol, ethyl alcohol, methyl ethyl ketone or the like may be used in combination in order to increase solubilities of an acid catalyst and a dye.

The dyes and/or pigments used in the invention are selected from those having good weatherability for the purpose of coloring a film. Examples of the dye include direct dyes such as C.I. Direct Yellow 98, C.I. Direct Red 220, C.I. Direct Blue 77 and the like, and acid dyes such as C.I. Acid Yellow 112, C.I. Acid Red 256, C. I. Acid Blue 182 and the like. Examples of the pigment include inorganic pigments such as C.I. Pigment Yellow 157, C.I. Pigment Red 101, C.I. Pigment Blue 29 and the like, and organic pigments such as C.I. Pigment Yellow 154, C.I. Pigment Red 122, C.I. Pigment Blue 15:1 and the like. These dyes and/or pigments may be used singly or in combination. Moreover, fluorescent pigments for imparting a fluorescent color, light-storing pigments of the type wherein afterglow continues to the extent after light irradiation has stopped, iridescent pigments for imparting iridescence, temperature-indicating pigments capable of color change depending on an ambient temperature, hydrophilic pigments for imparting hydrophilicity, functional pigments for reflecting an IR ray (heat ray) and the like may be appropriately used depending on the purpose. Mention is made of a pigment made of an acrylic resin dissolving Acid Yellow 73 therein as an example of the fluorescent pigment, a pigment composed mainly of strontium aluminate as an example of the light storing pigment, a pigment composed

mainly of titanium dioxide-coated mica as an example of the iridescent pigment, a pigment containing, in microcapsules, rhodamine B lactam/isooctyl galate/cetyl alcohol as an example of the temperature-indicating pigment, a pigment composed mainly of silica and titania as an example of the hydrophilic pigment, and ATO (antimony/tin oxides) ITO (indium/tin oxides) and the like fine powders as an example of the functional pigment for reflecting an IR ray (heat ray).

In the colored, transparent film-forming composition of the invention, it is preferred to further add (f) a stabilizer for an unpaired electron pair of a nitrogen atom and in order to inhibit the reaction with a UV absorber as will be caused by the unpaired electron pair of the nitrogen atom derived from an amino group-containing alkoxysilane having active hydrogen.

Various compounds may be mentioned as such a stabilizer (f), and salicylic acid, fumaric acid, crotonic acid, succinic acid, tartaric acid, para-hydroxybenzoic acid, pyrogallol, resorcinol and the like are preferably used.

These compounds used as (f) stabilizer may be used singly or in combination.

It is necessary that prior to coating onto a transparent substrate, an oil deposit or the like be removed from the surface of the transparent substrate such as of glass. Many methods of removing an oil film has been hitherto known, of which a method using a compound for oil film separation is preferred.

The colored, transparent film-forming composition of the invention can be applied ~~te~~by use of a brush, felt, non-woven fabric or the like.

For the coating, it is preferred to coat the composition in the direction of gravity so that ~~there~~the unevenness is unlikely to occur.

The colored, transparent film-forming composition of the invention is coated onto a transparent substrate, such as of glass, at room temperature, and drying to the touch is obtained within a range of from 0.5 hours to 2 hours. Further drying over 12 hours to 24 hours results in a transparent, hard, cured film of a beautiful color.

If the cured film, which is obtained by coating the colored, transparent film-forming composition of the invention onto a transparent substrate such as of glass, undergoes color degradation or is lowered in the effect of UV absorption, it is removed from the transparent substrate such as of glass and subjected to coating again.

In the practice of the invention, an alkaline detergent is used as a remover.

The alkaline agent should preferably have a pH of 9 or over. When the pH of the alkaline agent is smaller than 9, the solubility in the film is worsened. It will be noted that when the use of an alkaline agent alone does not permit infiltration into and cleaning of the film, an alkaline detergent is used wherein one or more nonionic surface active agents are formulated. Examples of the alkaline agent include anionic surface active agents, bicarbonates, sesqui-carbonates, carbonates, metasilicates, orthosilicates, phosphates, pyrophosphates, tripolyphosphates, metaborates, borates and the like in the form of sodium or potassium salts, and water-soluble amino alcohols such as triethanolamine, diethanolamine, monoethanolamine, methyldiethanolamine, ethyldiethanolamine, monoisopropanolamine, aqueous

ammonia and the like. One or more of these compounds are appropriately dissolved in water and adjusted in pH to a level of 9 or over.

In addition, abrasives may be used in combination, and it is convenient to softly rub the surface with a piece of abrasive-attached sponge.

The embodiments of the invention may be summarized as follows.

(1) A colored, transparent film-forming composition comprising (a) a reaction product of an epoxy group-containing alkoxysilane (a-1) and an amino group-containing alkoxysilane ~~(a-2)~~ having active hydrogen (a-2) therein, (b) an acid an acid catalyst, (c) an alkali-soluble UV absorber, (d) at least one solvent selected from organic solvents having a boiling point of 100 to 250°C, and (e) a dye and/or a pigment.

(2) A colored, transparent film-forming composition as recited in (1) above, wherein the reaction product is one which is obtained by reacting the epoxy group-containing alkoxysilane (a-1) and the amino group-containing alkoxysilane ~~(a-2)~~ having active hydrogen (a-2) at a ratio of by weight of 6:4 to 9:1.

(3) A colored, transparent film-forming composition as recited in (1) or (2) above, wherein the amino group-containing alkoxysilane ~~(a-2)~~ having active hydrogen (a-2) consists of N-(3-aminoethyl)-γ-aminopropyltrimethoxysilane.

(4) A colored, transparent film-forming composition as recited in any one of (1) to (3) above, wherein one or more of solvents selected from alcohol solvents, ketone solvents, ether solvents and solvents having two or more functional groups are used as an essential component.

(5) A colored, transparent film-forming composition as recited in any one of (1) to (4) above, further comprising, as (f) a stabilizer, one or more of organic compounds selected from salicylic acid, fumaric acid, crotonic acid, succinic acid, tartaric acid, parahydroxybenzoic acid, pyrogallol and resorcinol.

(6) A colored, transparent film-forming composition as recited in any one of (1) to (5) above, wherein the alkali-soluble UV absorber is used in an amount of 5 to 40 wt% of the film-forming component.

(7) A method for coating a colored, transparent film-forming composition as recited in any one of (1) to (6) above, which comprises storing (e) the dye and/or pigment component of the composition in a separate container, mixing the other components with the dye and/or pigment component immediately before the use of the composition to provide a mixed solution, impregnating the mixed solution in a coating applicator, and applying onto a pretreated transparent substrate in a direction of gravity.

(8) A method for removing a cured film comprising separating, after use, a colored film formed by curing a colored, transparent film-forming composition as recited in (1) to (6) above on a transparent substrate from the transparent substrate by use of an alkaline detergent.

The invention is described by way of examples.

#### Example A

Preparation of a colored transparent film-forming composition

(A-1)

(1) 85 g of  $\gamma$ -glycidoxypropyltrimethoxysilane and 15 g of  $\gamma$ -aminopropyltriethoxysilane were mixed and



agitated for 1 hour, followed by allowing to stand in a thermostatic chamber at 25°C for 14 hours and thus aging, thereby obtaining 100 g of reaction product (I).

(2) 5 g of 2,4-dihydroxybenzophenone was dissolved in 20 g of propylene glycol monoethyl ether acetate.

(3) 0.2 g of boron trifluoride piperidine was dissolved in 10 g of dipropylene glycol monomethyl ether propanol.

(4) The solution of (3) was mixed with the solution of (2), followed by further uniform mixing of 20 g of the solution of (1) to prepare solution A-1.

(A-2)

(5) 65 g of  $\gamma$ -glycidoxypentyltrimethoxysilane and 35 g of  $\gamma$ -aminopentyltriethoxysilane were mixed, followed by agitation and aging in the same manner as in (1) to obtain 100 g of reaction product (II).

(6) The solutions of (2) and (3) above were, respectively, prepared in the same manner as set out above, and the solution of (3) was mixed with the solution of (2), followed by further uniform mixing of 20 g of the solution of (5) to prepare solution A-2.

(A-3)

(7) 75 g of  $\gamma$ -glycidoxypentyltrimethoxysilane and 25 g of  $\gamma$ -aminopentyltriethoxysilane were mixed, followed by agitation and aging in the same manner as in (1) to obtain 100 g of reaction product (III).

(8) 5 g of 2,2', 4,4'-tetrahydroxybenzophenone was dissolved in 20 g of propylene glycol monoethyl ether acetate.

(9) The solution of (3) was prepared in the same manner as set out above and mixed with the solution of (8), followed by further uniform mixing of 20 g of the solution of (7) to prepare solution A-3.

(A-4)

(10) 75 g of  $\gamma$ -glycidoxypropyltrimethoxysilane and 25 g of  $\gamma$ -(2-aminoethyl)aminopropyltrimethoxysilane were mixed, followed by agitation and aging in the same manner as in (1) to obtain 100 g of reaction product (IV).

(11) The solution of (3) was prepared in a similar manner and mixed with the solution of (8), followed by further uniform mixing of 20 g of the solution of (1) to prepare solution A-4.

(A-5)

(12) The solutions of (2) and (3) were similarly prepared, and the solution of (3) was mixed with the solution of (2), followed by complete dissolution of 5 g of salicylic acid and further uniform mixing of 20 g of the solution of (5) to prepare solution A-5.

#### Comparative Example

(13) 5 g of 2-hydroxy-4-methoxybenzophenone was dissolved in 20 g of propylene glycol monocethyl ether acetate.

(14) The solution of (3) was similarly prepared and mixed with the solution of (13), followed by uniform mixing of 20 g of the solution of (5) to prepare a reference solution.

Next, C.I. Pigment Yellow 154 was dispersed in dipropylene glycol monomethyl ether propanol in an amount of 1 wt% to obtain a colored solution. The colored solution was mixed with the solutions A-1 to A-5 and the reference solution at a ratio by weight of 1:1, respectively, thereby preparing colored, transparent film-forming compositions.

Table 1

		Component (a)	Component (b)	Component (c)	Component (d)	Component (e)	Component (f)
Example	A-1	Reaction product (I)	BF <sub>3</sub> piperidine	2,4- dihydroxy- benzo- phenone	Dipropylene glycolmonomethyl ether propanol/propylene glycol monoethyl ether acetate	C.I. Pigment Yellow 154	Not added
	A-2	Reaction product (II)	BF <sub>3</sub> piperidine	2,4- dihydroxy- benzo- phenone	Dipropylene glycolmonomethyl ether propanol/propylene glycol monoethyl ether acetate	C.I. Pigment Yellow 154	Not added
	A-3	Reaction product (III)	BF <sub>3</sub> piperidine	2,2',4,4'- tetra- hydroxy- benzo- phenone	Dipropylene glycolmonomethyl ether propanol/propylene glycol monoethyl ether acetate	C.I. Pigment Yellow 154	Not added
	A-4	Reaction product (IV)	BF <sub>3</sub> piperidine	2,2',4,4'- tetra- hydroxy- benzo- phenone	Dipropylene glycolmonomethyl ether propanol/propylene glycol monoethyl ether acetate	C.I. Pigment Yellow 154	Not added
	A-5	Reaction product (II)	BF <sub>3</sub> piperidine	2,4- dihydroxy- benzo- phenone	Dipropylene glycolmonomethyl ether propanol/propylene glycol monoethyl ether acetate	C.I. Pigment Yellow 154	Salicylic acid
Reference		Reaction product (II)	BF <sub>3</sub> piperidine	2,4- dihydroxy- benzo- phenone	Dipropylene glycolmonomethyl ether propanol/propylene glycol monoethyl ether acetate	C.I. Pigment Yellow 154	Not added

Example B

\_\_\_\_\_ Coating of colored transparent film-forming  
 compounds and curing of films

An example of coating on window glass of an  
 automobile is shown.

(Step of removing an oil film)

An oil film-removing compound is attached to a  
 polishing sponge soaked with a small amount of water,  
 with which the entire surface of the window glass ~~are~~is  
 thoroughly polished. In order to confirm whether or not  
 the oil film is completely removed, the glass is wholly  
 wiped with the water-soaked sponge. Where water ~~is~~does  
 not attached to the glass surface in the form of

droplets, the oil film is left, so that the above procedure using the oil film-removing compound is repeated until the glass surface is fully wetted with water. After complete wiping-off of the water and compound, the window is entirely de-fatted with a non-woven fabric piece folded up several times and soaked with isopropyl alcohol.

(Coating step of colored, transparent film-forming compositions)

About 8 ml of a solution of each colored, transparent film-forming composition prepared in Example A was placed in a tray (50 x 70 mm in width x 15 mm in depth) having a capacity of 30 ml, and all the liquid in the tray was soaked in a piece of non-woven fabric folded up several times (5 x 30 mm in width x 20 mm in thickness) at a coating side thereof (50 mm in width).

The holding side of the non-woven fabric piece (i.e. a side opposite to the coating side) is firmly held, after which the liquid soaked in the piece is gently coated down in the form of a band from the upper end at the right or left end of the window toward the lower end (in a direction of gravity). After arrival at the lower end, the procedure of coating in the form of a strip from the upper to lower end (in a direction of gravity) is repeated so that the strips are superposed by about 1/3 to 1/4 thereof, thereby uniformly coating the whole of the window.

(Drying step)

After the coating, the window is allowed to stand in a place where water is not splashed on and dust is reduced in amount and is thus naturally dried. Usually, it takes 30 minutes in summer and 2 hours or below in

winter before drying to such an extent that the liquid is not attached to the finger on contact with the coated surface (drying to the touch). Thereafter, the drying is continued by allowing to stand over about 24 hours to ensure substantially a complete drying state wherein the film is cured to an extent of suffering little~~few~~ defects when the window is opened or closed.

(Dry-to-the-touch time)

Based on the method of JIS K 5400, the dry-to-the-touch time was measured. (Measured at intervals of 10 minutes at 25°C in summer and at 10°C in winter.)

After the coating, the film state after passage over 72 hours was assessed in the following manner.

(Transparency)

Assessed through visual observation based on the method of JIS K 5400.

(Unevenness of coating)

Assessed through visual observation based on the method of JIS K 5400.

(Film hardness)

The film hardness was assessed according to a pencil scratching method based on JIS K 5400.

Each colored, transparent film-forming composition prepared in Example A was coated and dried on a glass test sheet (70 mm x 110 mm in width x 5 mm in thickness) in the same manner as in the procedure of coating on the automotive window glass. After the coating, the test sheet was naturally dried over 24 hours and subjected to measurement of a UV transmittance in the following manner.

(UV transmittance)

The measurement was conducted by use of a spectrophotometer wherein a UV transmittance at a wavelength of 345 nm was evaluated.

Moreover, the test sheet was set in an accelerated light fastness tester (prescribed in JIS B 7754), and was again subjected to similar measurement of UV transmittance after passage of 192 hours.

It was revealed that the compositions of Examples A-1 to A-5 exhibited initial UV transmittances lower than the reference, with the UV transmittance being low after the accelerated light fastness test, thus ensuring an excellent UV-absorbing effect which continued over a long time.

These results are shown in Table 2 below.

		Dry-to-the-touch time	Trans- parency	Uneveness of Coating	Film hardness	UV trans- mittance (after 24 hours)	UV trans- mittance ( 192 hours after weatherability test)
Examp- le	A-1	Summer (25°C) 40 minutes	good	no	H	5%	20%
		Winter (10°C) 90 minutes	good	no	H	5%	20%
	A-2	Summer (25°C) 30 minutes	good	no	2H	10%	35%
		Winter (10°C) 80 minutes	good	no	2H	10%	35%
	A-3	Summer (25°C) 30 minutes	good	no	2H	0%	10%
		Winter (10°C) 80 minutes	good	no	2H	0%	10%
	A-4	Summer (25°C) 30 minutes	good	no	3H	0%	15%
		Winter (10°C) 80 minutes	good	no	3H	0%	15%
	A-5	Summer (25°C) 40 minutes	good	no	2H	0%	10%
		Winter (10°C) 90 minutes	good	no	2H	0%	10%
Reference		Summer (25°C) 30 minutes	good	no	2H	30%	90%
		Winter (10°C) 80 minutes	good	no	2H	30%	90%

### Example C

#### Removal of cured film

The cured film obtained by coating each colored, transparent film-forming composition prepared in Example A on the window glass of the actual car in Example B was subjected to outdoor exposure over 3 months (June to August in summer and December to February in winter). Thereafter, the film surface was softly rubbed with a sponge piece soaked with an anionic surface active agent (i.e. an oleic acid triethanolamine aqueous solution/pH = 9) provided as alkali detergent I, and the removability of the film of each sample was assessed on the basis of the following standards.

Further, an alkaline abrasive composition (pH = 10, pasty composition) provided as alkali detergent II wherein an anionic surface active agent (oleic acid morpholine) was used, in which a petroleum solvent and calcined diatomaceous earth serving as an abrasive are, respectively, emulsified and dispersed, was prepared, and the film surface was softly rubbed with a sponge piece

@: excellent (quickly, completely removed)

○: good (completely removed)

△: slightly bad (not removed partially)

×: bad (not removed)

It has been found that Examples A-1 to A-5 were superior in removability of the cured films to the reference. The results are shown in Table 3.

Table 3

		Period of outdoor exposure	Alkali detergent I	Alkali detergent II
Example	A-1	3 months in summer	○ - △	○
		3 months in winter	○ - △	○
	A-2	3 months in summer	○ - △	○
		3 months in winter	○ - △	○
	A-3	3 months in summer	○	⊙
		3 months in winter	○	⊙
	A-4	3 months in summer	○	⊙
		3 months in winter	○	⊙
	A-5	3 months in summer	⊙	⊙
		3 months in winter	⊙	⊙
Reference		3 months in summer	×	×
		3 months in winter	×	×

As will be seen from the results of Tables 2 and 3, according to the invention, there can be obtained a colored, transparent film-forming composition that is free of unevenness of coating and color shading and can be cured at room temperature and that can form an attractive, transparent, UV-absorbing, colored film having a high film hardness after curing. In addition, after use, the film can be readily removed in a simple manner.



IN THE CLAIMS

Please amend Claims 1-3 as follows.

1. (Amended) A colored, transparent film-forming composition comprising (a) a reaction product of an epoxy group-containing alkoxysilane (a-1) and an amino group-containing alkoxysilane ~~(a-2)~~ having active hydrogen therein (a-2), (b) an acid catalyst, (c) an alkali-soluble UV absorber, (d) at least one ~~solvent selected from~~ organic solvents having a boiling point of 100 to 250°C, and (e) a dye and/or a pigment.

2. (Amended) A colored, transparent film-forming composition according to Claim 1, wherein said reaction product is one obtained by reaction between the epoxy group-containing alkoxysilane (a-1) and the amino group-containing alkoxysilane ~~(a-2)~~ having active hydrogen (a-2) at a ratio by weight of 6:4 to 9:1.

3. (Amended) A colored, transparent film-forming composition according to Claim 1, wherein the amino group-containing alkoxysilane ~~(a-2)~~ having active hydrogen (a-2) consists of N-( $\beta$ -aminoethyl)- $\gamma$ -aminopropyltrimethoxysilane.